Radiation Safety Course

Review Session

Units & Doses 1.Prefixes a. 6 cents $x = \frac{1 \text{ }}{100 \text{ cents}} = \frac{0.06 \text{ }}{100 \text{ cents}}$ 0.06 \$ $x = \frac{1,000 \text{ m}}{100 \text{ cents}} = \frac{60 \text{ m}}{100 \text{ cents}}$ b. 100 \$\mu \text{ci} \text{ } \frac{1 \text{ Ci}}{1 \text{ } 10^6 \text{ }} = \frac{1 \text{ } x \text{ } 10^{-4} \text{ Ci}}{1 \text{ } 10^6 \text{ }} \text{ c. 3.3 Ci} \text{ } \frac{1,000 \text{ mCi}}{1 \text{ Ci}} = \frac{3,300 \text{ mCi}}{1 \text{ } 10^6 \text{ }} \text{ d. 2 } \text{ } \frac{1 \text{ Ci}}{1 \text{ } x \text{ } 10^6 \text{ }} = \frac{2 \text{ Ci}}{1 \text{ } x \text{ } 10^6 \text{ }} \text{ Ci}

Radiation Safety Review

- 1. Prefixes, continued
 - e. 50 mrem x $\frac{1 \text{ rem}}{1,000 \text{ mrem}} = 0.05 \text{ rem}$
 - f. 42 R/h x $\frac{1,000 \text{ mR/h}}{1 \text{ R/h}} = 42,000 \text{ mR/h}$
 - g. $10^6 \text{ dpm x } \frac{1\mu\text{Ci}}{2.22 \text{ x } 10^6 \text{ dpm}} = 0.45 \mu\text{Ci}$
 - h. $4 \mu \text{Ci } \times \frac{3.7 \times 10^4 \text{ dps}}{1 \mu \text{Ci}} = 1.48 \times 10^5 \text{ dps}$

Da Lindian Calata Danian

- 2. Units: Indicate the correct units:
- (a) MPD for eyes of a radiation worker is 15

 $\underline{\hspace{1cm}}$ mrem/y $\underline{\hspace{1cm}}$ rem/y $\underline{\hspace{1cm}}$ R/y

(b) Effective whole body dose equivalent from naturally occuring radiation sources is 300

 $\underline{\hspace{1cm}}$ $\mu \text{Ci/y}$ $\underline{\hspace{1cm}}$ rem/y $\underline{\hspace{1cm}}$ mrem/y

- (c) The absorbed dose rate from an activity of 1 μCi per cm² of ³²P deposited on the skin is 9
 - X rads/h rads/cm² μ Ci/cm²

Radiation Safety Review

3. Radiation Exposures: Absorbed Doses & Dose Equivalents

Select answers from: 0 mR, 20 mR, 5R/h, 0 mrads, 100 mrads, 340 rads, 0 mrem, 5 mrem, 1250 mrem, 1 rem, 1.25 rem, 20 rem, 50 rem

- a. Maximum permissible <u>dose equivalent</u> (gonads) radiation worker: <u>50 rem/y, 20 rem/year -implied</u>
- b. "Radiation Area" = whole body <u>dose equivalent</u> ≥ 5 mrem/hour.
- c. Whole body <u>dose equivalents</u> which gives cancer mortality risk of 4 in 10,000: <u>1 rem</u>

Radiation Safety Review

- 3. Radiation exposures, cont.
- d. LD₅₀ <u>absorbed dose</u> (with minimal treatment): 340 rads
- e. Exposure rate at 1 cm from 10 mCi of ⁸⁶Rb: <u>5 R/hour</u>.
- f. <u>Dose equivalent</u> from 1 rad of alpha radiation: <u>20 rem</u>.
- g. <u>Absorbed dose</u> threshold for induction of genetic effects: <u>0 mrads</u>.

4. What is the total dose equivalent which should be recorded in a person's dosimetry records if he/she has received the following individual doses? (Give answer in conventional and SI units.)

x-rays: 210 mrads x 1 = 210 mrem x 20 = 1200 mrem alpha rays: 60 mrads beta rays: 150 mrads x 1 = 150 mrem 30 mrads x 1 gamma rays: 30 mrem thermal neutrons: 10 mrads 23 mrem TOTAL: 1,613 mrem

 $1,613 \text{ mrem} \quad x \quad \frac{1 \text{ rem}}{1,000 \text{ mrem}} = 1.613 \text{ rem}$

1.613 rem x $\frac{10 \text{ mSv}}{1 \text{ rem}}$ = 16.13 mSv

Radiation Safety Review

5. SI Units:

a. 1 mGy
$$\times$$
 $\frac{100 \text{ mrads}}{1 \text{ mGy}} = 100 \text{ mrads}$

b.
$$4 \mu Ci$$
 $x = \frac{3.7 \times 10^4}{1 \mu Ci} Bq = 1.48 \times 10^5 Bq$

c. 74 kBq x
$$\frac{1,000 \text{ Bq}}{1 \text{ kBq}}$$
 x $\frac{1 \mu \text{Ci}}{3.7 \text{ x } 10^4 \text{ Bq}} = 2 \mu \text{Ci}$

d. 5 mrads/h
$$\times \frac{0.01 \text{ mGy/h}}{1 \text{ mrad/h}} = 0.05 \text{ mGy/h}$$

Radiation Safety Review

- 5. SI Units, cont.
- e. 1.25 rem x $\frac{1,000 \text{ mrem}}{1 \text{ rem}} = 1,250 \text{ mrem}$

 $1.25 \text{ rem } \times \frac{10 \text{ mSv}}{1 \text{ rem}} = 12.5 \text{ mSv}$

 $1.25 \text{ rem } \times \frac{0.01 \text{ Sy}}{1 \text{ rem}} = 0.0125 \text{ Sy}$

f. 5 mCi x $\frac{37 \text{ Mbq}}{1 \text{ mCi}}$ = 185 Mbq

g. 10 rads $\frac{1 \text{ cGy}}{1 \text{ rad}} = 10 \text{ cGy}$

Radiation Safety Review

Problem Set: Basics of Radiation Safety

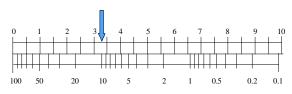
- 1. How much ¹²⁵I activity remains 200 days after the calibration date if the initial activity is 10 mCi?
- A. Nomogram method Half-life of $^{125}I = 60$ days # of half-lives = 200 days/60 days = 3.33

From nomogram (p. A11), 10% of the initial activity remains, so: $A = 0.1 \times 10 \text{ mCi} = 1 \text{ mCi}$

Radiation Safety Review

Decay Nomogram

Number of Half-lives



Percentage Remaining

Radiation Safety Review

- 1. cont.
- B. Formula method Half-life of $^{125}I = 60$ days

$$\begin{array}{lll} A = & A_0 e^{(-0.693)(t/Tl/2)} & = & 10e^{(-0.693)(200/60)} \\ = & 10^{-2.31} & = & 10 & x & 0.099 & = & 0.99 \ mCi \end{array}$$

(or)

$$\begin{array}{lll} A = \ A_0 (1/2)^N = \ 10 (1/2)^{(200/60)} & = \ 10 (1/2)^{3.33} \\ = \ 10 & x \ 0.099 \ = \ 0.99 \ mCi \end{array}$$

- 2. A certain procedure which involves the use of ¹⁹⁸Au is performed over a 4 hour period at a working distance of 20 cm. If the exposure rate from the ¹⁹⁸Au is 10 mR/hr, the total integrated exposure to the radiation worker would be 40 mR. Assume that it is possible to change the working parameters of the procedure in order to reduce the exposure. What would the exposure be if
- a. The **time** were reduced to 2 hours? 20 mR (Time and therefore dose are cut in half.)
- b. The **distance** were increased to 60 cm? $I_2 = I_1(d_1/d_2)^2$ $I_2 = 40 \text{ mR}(20/60)^2 = 4.4 \text{ mR}$
- c. If 1 cm of lead **shielding** were used? (TVL = 10 mm Pb) $I = I_0 e^{(-2.3x/TVL)} = 40 \text{ mR} e^{(-2.3)10\text{mm}/10\text{mm})} = 4 \text{ mR}$
- d. If **all three of the above** were done? 40 mR x 20/40 x 4.4/40 x 4/40 = <u>0.22 mR</u>

Radiation Safety Review

3. What is the dose rate through the protective layer of skin from a 2 μ C i deposition of ³²P? (See 'Rules of Thumb for Beta Particles' on page A10.)

$$D(rad/h) = 9 B = 9 x \frac{2 \mu Ci}{cm^2} = \frac{18 \text{ rads/h}}{}$$

Radiation Safety Review

4. What is the exposure rate (gamma) at 20 cm from a 3 mCi vial of ¹³¹I?

 $\Gamma = 2.1 \text{ R/h} @ 1 \text{ cm per mCi (From Table)}$

The dose rate for 3 mCi: 3 x 2.1 = 6.3 R/h @ 1 cm

Using the inverse square law (page A11):

$$I_2 = I_1(d_1/d_2)^2 = 6.3(1/20)^2$$

= 6.3(1/400) = 0.0158 R/h = 15.8 mR/h

Radiation Safety Revie

5. If the exposure rate from some ⁵¹Cr with no shielding is 20 mR/hr, what is the exposure rate reduced to if 2 mm of Pb is used for shielding? (The TVL is 6.3 mm of Pb.)

A. Nomogram Method # of TVL's = 2/6.3 = 0.32From the nomogram on p. A10, 48% is remaining. I = 20 R/h x 0.48 = 9.6 mR/h

B. Formula Method

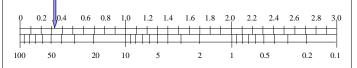
 $I = I_0 e^{(-2.3 \text{x/VL})}$ = (20 mR/h) $e^{(-2.3)(2/6.3)}$ = 20(0.482)= 9.6 mR/h
(or)

 $I = I_0(1/10)^n$, where n = # of TVL's = $(20 \text{ mR/h})(1/10)^{0.32} = (20 \text{ mR/h})(0.48)$ = 9.6 mR/h

Radiation Safety Review

TVL Nomogram

Number of TVL's



% Exposure Remaining

Radiation Safety Review

- 6. What is the meaning of the following terms?
 - a. LD₅₀

50% of the exposed population is expected to die within 60 days.

b. Genetic effects of radiation

Effects passed on to offspring.

7. A person who has an annual intake of a radioisotope equal to the ALI (Annual Limit on Intake) will receive an absorbed dose equal to the MPD (maximum permissible dose). What fraction of the MPD is received if a person has an intake of 1 mCi for the following isotopes?

<u>Isotope</u>	ALI (μCi)	Fraction or multiple of MPD
$^{3}\mathrm{H}^{^{2}}$	80,000	0.013
^{125}I	40	25
⁴⁵ Ca	800	1.25
14 C	2,000	0.5
⁹⁹ mTc	80,000	0.013

On the basis of the above numbers, which isotopes most toxic? ^{125}I , which is least toxic? ^{3}H , ^{99m}Tc

Radiation Safety Review

8. All of the following biological effects may result from exposure to radiation. Most, however, require large doses of radiation (there is a threshold below which the effect does not occur). Indicate those for which there is no known threshold, ie., may have a possibility of occurring even with low doses.

X mutations	nausea	X leukemia
_ sterility	_ dermal necrosi	s X cancer
erythema	_acute radiation sy	ndrome
$\underline{\mathbf{X}}$ genetic effects	epilation	

Radiation Safety Review

9. The radiological half-life of ¹²⁵I is 60 days. The biological half-life is 138 days (for soluble forms). What is the effective half-life?

$$T_{\text{eff}} = (T_{1/2}T_b)/(T_{1/2} + T_b)$$

= $(60)(138)/(60 + 138) = 41.8 \text{ days}$

Radiation Safety Review

10. Cancer induction by ionizing radiation(p. A18)

What is the threshold for cancer induction? 0 rads

What is the latent period for solid cancers? 10 to 35 years

What is the risk factor? 4/10,000 fatal cancers/rem of whole body dose (0.0004)

What is the risk of having a fatal cancer induced by a whole body radiation dose of 5 rem? 1 chance in 500

$$5 \text{ rem } \times \frac{4}{10,000} = \frac{20}{10,000} = \frac{1}{500}$$

Radiation Safety Review

11. What is the major risk to the embryo or fetus during the:

first 8 days of pregnancy C remainder of the first trimester A last 6 or 7 months of pregnancy D

- (a) birth defects
- (b) fetal death
- (c) death of embryo
- (d) increased risk of childhood cancer